

Construction of a Low Temperature Plasma Probe Capable of Ambient Ionization of Samples for Mass Spectrometric Analysis

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Introduction

- Ionization of Mass Spec Samples often use plasma sources.
- Same plasma sources may be large and costly.
- This method builds off of a previous model for a low temperature plasma probe.
- The probe is handheld and uses an AC square wave.¹
- The first probe was build using 125 kHz using a step up transformer to produce a dielectric barrier plasma.
- The second probe was build to power a piezoelectric transformer to generate a piezoelectric plasma, using a 20 V power supply.

Methods

Components used in circuit:

- 555-Timer
- N-channel MOSFET
- Transformer
- 9V Battery
- Resistors
- Capacitors
- Potentiometer

The 555-timer's frequency is variable based on how it is wired.

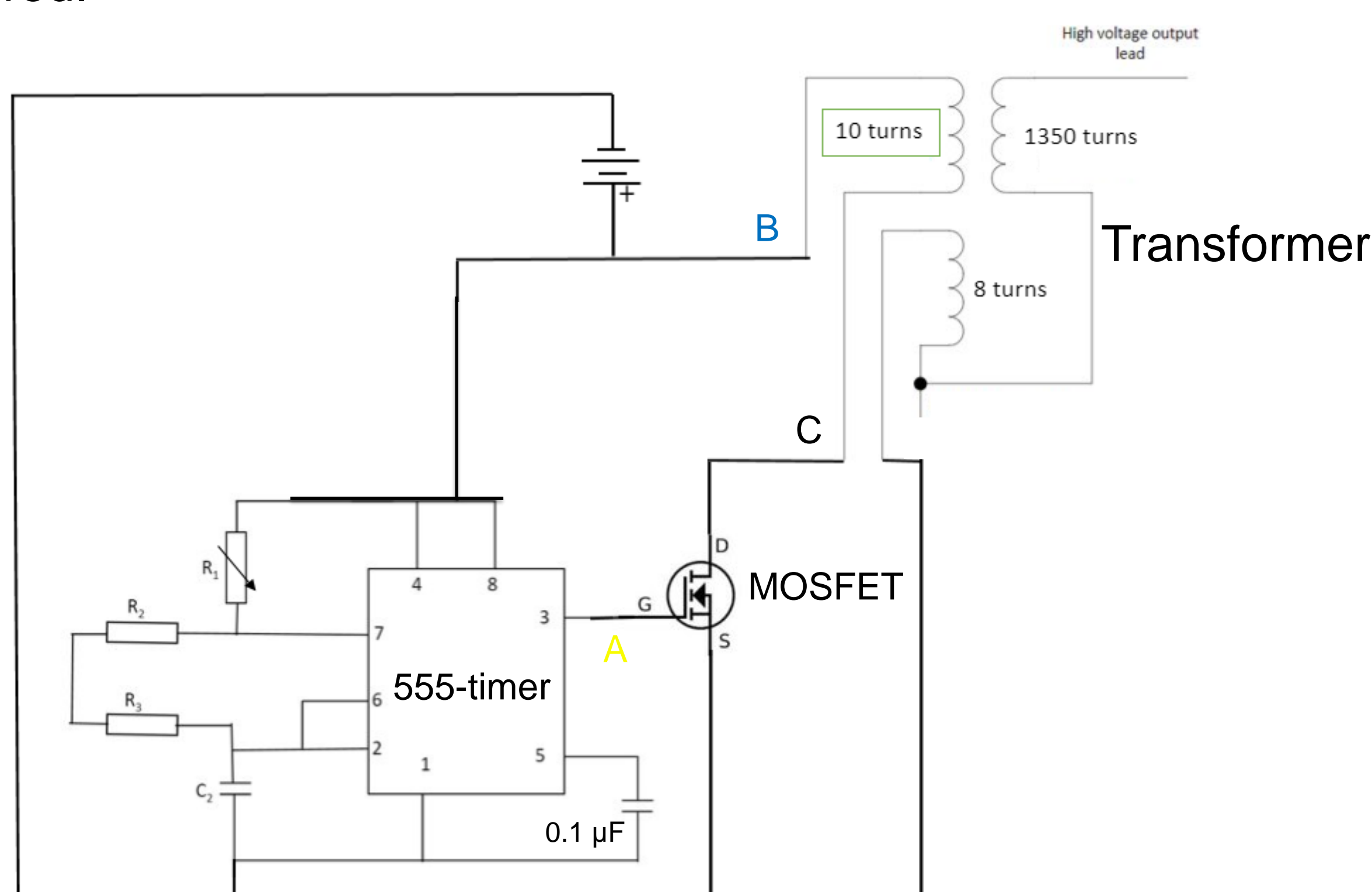


Figure 1. A diagram of the circuit used. A 2 kΩ pull-down resistor from the gate of the MOSFET to ground is not shown

Equations Used to Determine the Output of the 555-Timer

$$\text{Time High} = T_1 = 0.693(R_1 + R_2)(C)$$

$$\text{Time Low} = T_2 = 0.693(R_2)(C)$$

$$\text{Frequency} = f = 1/T = 1.44/(R_1 + 2R_2)(C)$$

$$\text{Duty Cycle} = T_{\text{ON}} / (T_{\text{ON}} + T_{\text{OFF}}) = ((R_1 + R_2) / (R_1 + R_2)) \%$$

Results: Dielectric Barrier Plasma

When the circuit was designed to generate a dielectric barrier discharge plasma, a plasma was achieved in ambient temperature under the following conditions:

$$R_1 = 1\text{ k}\Omega$$

$$R_2 + R_3 = 10\text{ k}\Omega$$

$$C_2 = 470\text{ pF}$$

Output signal from test point A (Figure 4):

$$\text{Frequency} = 110\text{ kHz}$$

$$\text{Duty Cycle} = 55.07\%$$

$$V_{\text{out}} = 9\text{ V}$$

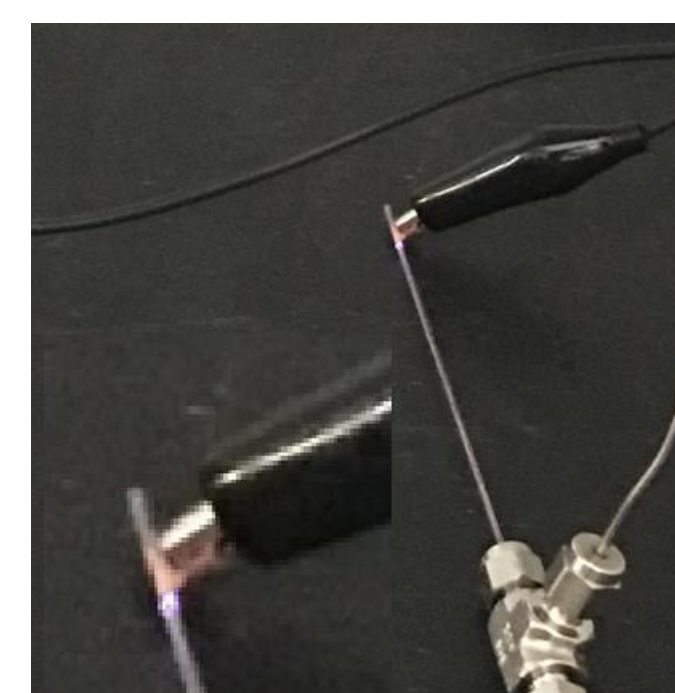


Figure 2. The dielectric barrier plasma generated in ambient conditions.

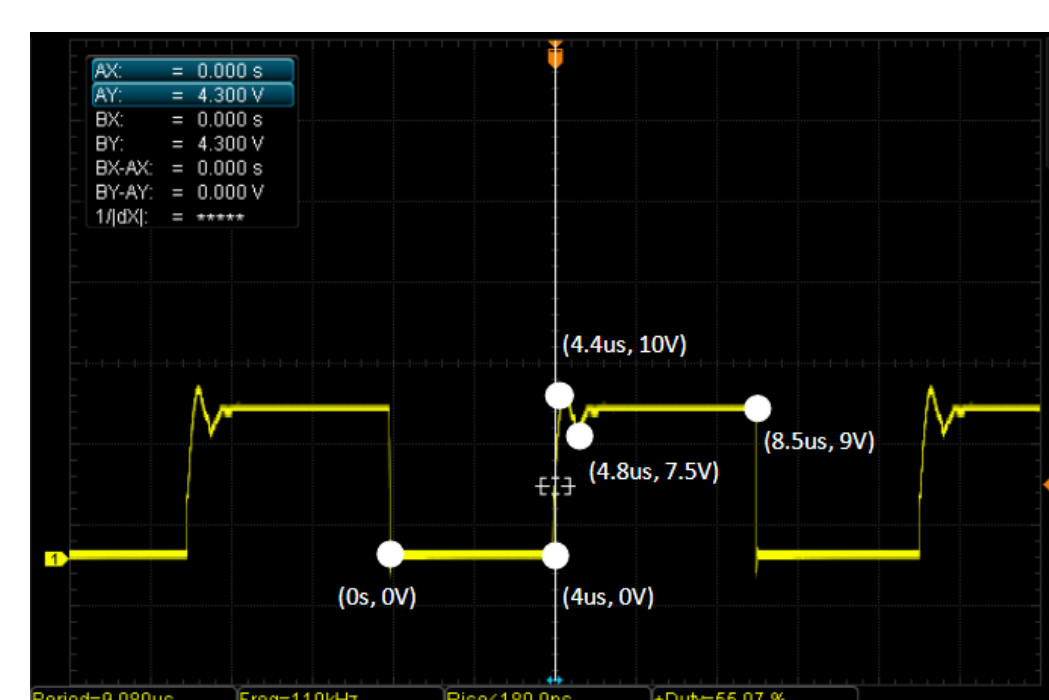


Figure 4. Oscilloscope data depicting the frequency and duty cycle of the output of the 555-timer

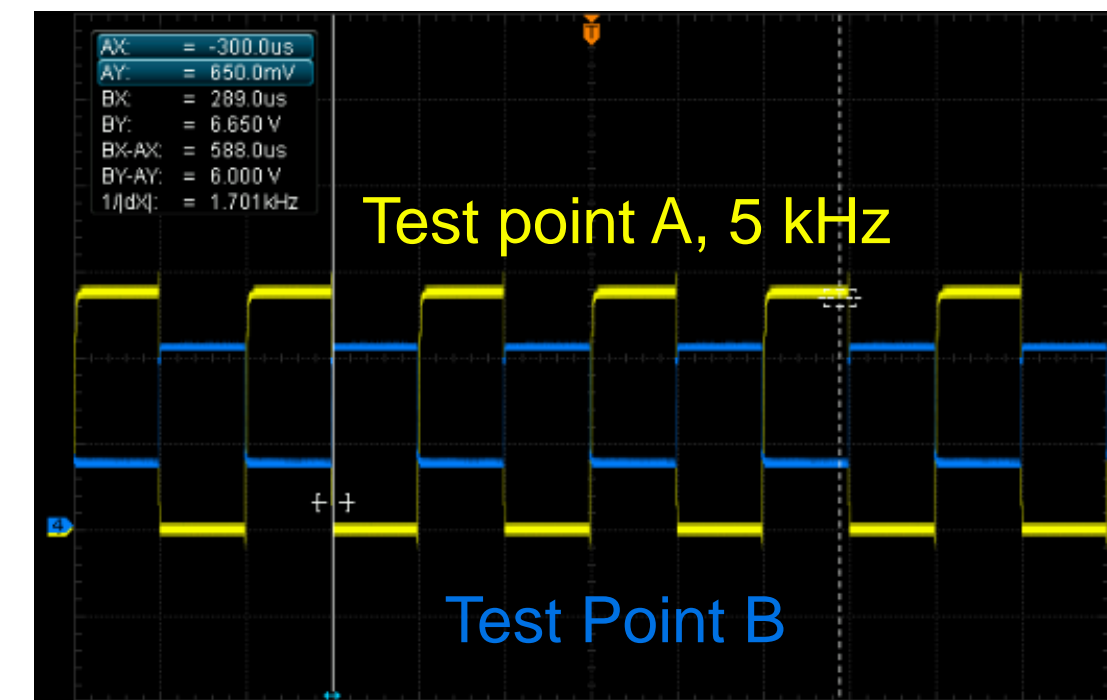


Figure 5: The signal measured from the output of the 555-Timer (yellow) and the signal measured from the N-Channel MOSFET (blue) to show signal inversion

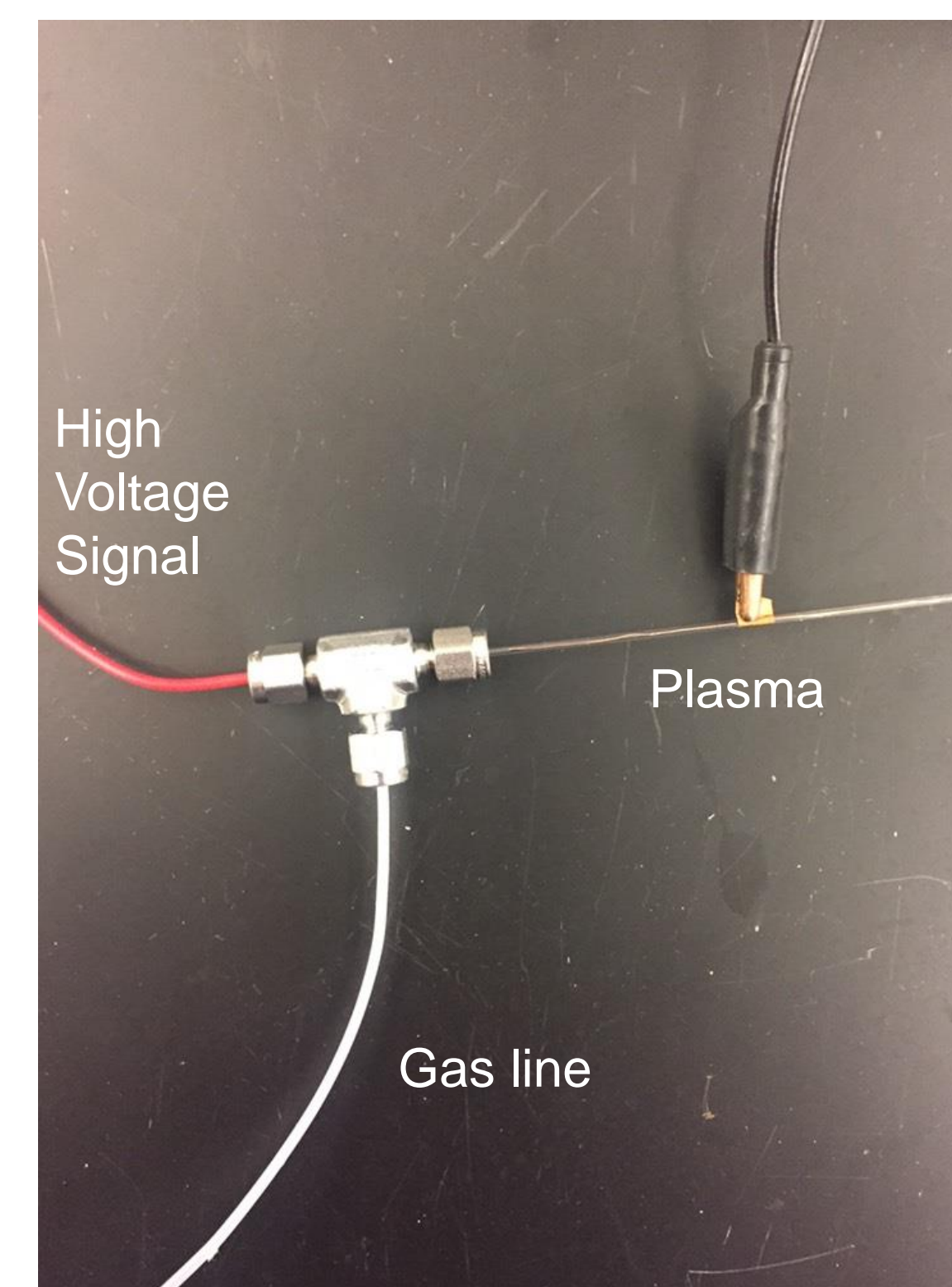


Figure 3. A labeled diagram of the plasma probe

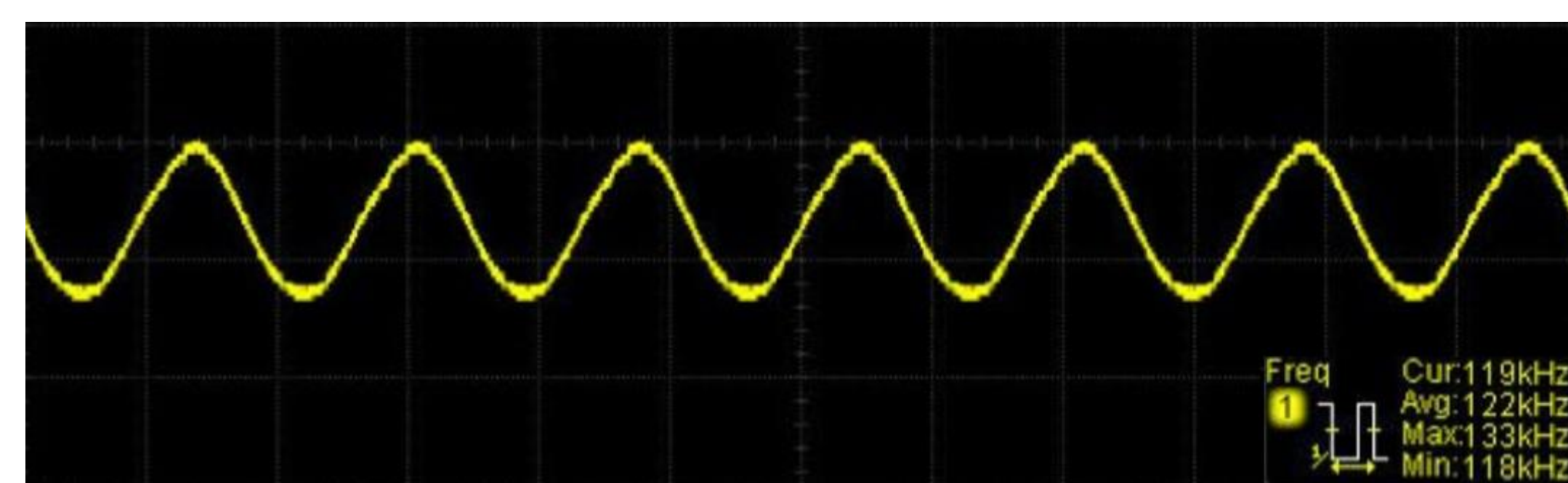


Figure 6. The resultant waveform from test point C. The signal is 119 kHz, 1.35 kV, the sine wave properties are unexpected but work for plasma generation.

Results: Piezoelectric Plasma

Plasma generation has not yet been achieved. The following conditions were used:

$$R_1 = \text{Potentiometer ranging from } 7\text{ k}\Omega \text{ to } 75\text{ k}\Omega$$

$$R_2 + R_3 = 90\text{ k}\Omega$$

$$C_2 = 100\text{ pF}$$

Output signal from test point A (Figure 4):

$$\text{Frequency with } R_1 \text{ as } 7\text{ k}\Omega = 55\text{ kHz}$$

$$\text{Frequency with } R_1 \text{ as } 75\text{ k}\Omega = 77\text{ kHz}$$

$$\text{Duty Cycle} = 64\%$$

$$V_{\text{out}} = 9\text{ V}$$

Same waveform as when the dielectric barrier plasma is used

Output signal from test point B:

20 V DC signal

Discussion

- Dielectric barrier plasma generation was successful
- Piezoelectric transformer's signal (test point B) is a DC signal.
- The 555-timer's signal driving the MOSFET for the piezoelectric transformer worked as expected.
- Piezoelectric transformers resistance should be tested, or equivalent resistor should be used in place for testing.
- When small resistor (i.e. 10 kΩ) is used, test point A performs as test point B with inversion of signal.

Summary and Conclusions

- Piezoelectric transformer deviates from expected functionality.
- Dielectric barrier plasma is achievable under ambient conditions with air.
- Signal differs when power supply vs. 9v battery is used.
- Potentiometer is capable of deviating the frequency by a large amount.

Future Work

- Continue diagnostic testing on the piezoelectric transformer.
- Complete the piezoelectric transformer.
- Use inert gas with the dielectric barrier plasma
- Collect Mass Spec data on various samples using these plasma probes.
- Compare MS results using these probes to other ionization methods.

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References

1. Wiley, J., Shelley, J., Cooks, R.; 2013. *Anal. Chem.* 85, 6545-6552