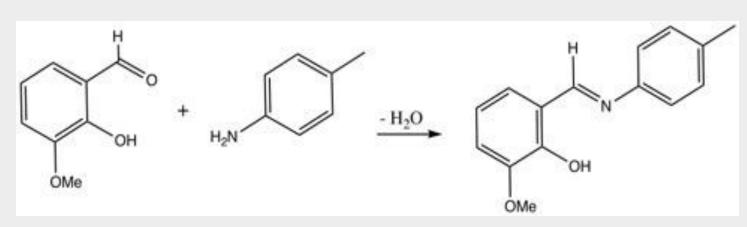
Optimization of N-(2-hydroxy-3-methoxybenzyl)-N-p-tolylacetamide via Reductive Amination

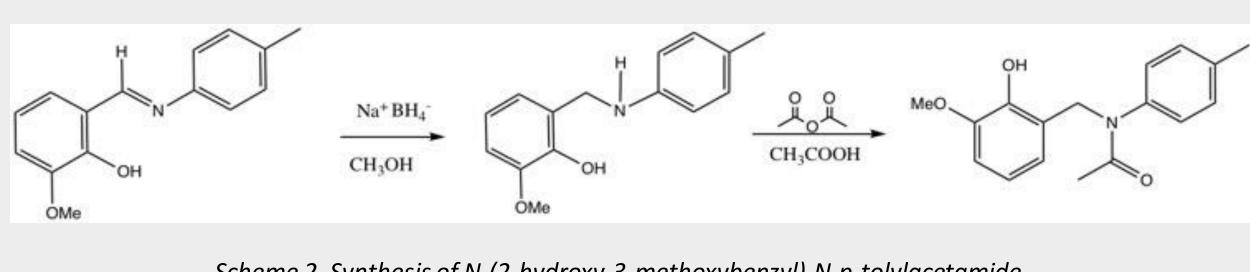
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Introduction

- The synthesis of imines, amines and amides is fundamental to our understanding of biological pathways.¹
- Several naturally occurring amines such as serotonin and epinephrine function as neurotransmitters in the body.
- Formation of amide bonds is one of the most important reactions in chemistry due to the pharmaceutical industry.
- An acetamide is an enanthic acid with an amide attached.²
- Acetamides are commonly used in syntheses as plasticizers
- Amination is the process of adding an amine group to a molecule and can be a powerful method of reducing functional groups.
- Reductive amination reduces imine to an amine.
- Proton and carbon-13 NMR are used to analyze the products and make preliminary judgements on the structure of the molecule.³
- IR spectroscopy is used to determine functional groups that would further classify the molecule.⁴



Scheme 1. Synthesis of 2-methoxy-6-(p-tolyliminomethyl)-phenol.



Scheme 2. Synthesis of N-(2-hydroxy-3-methoxybenzyl)-N-p-tolylacetamide.

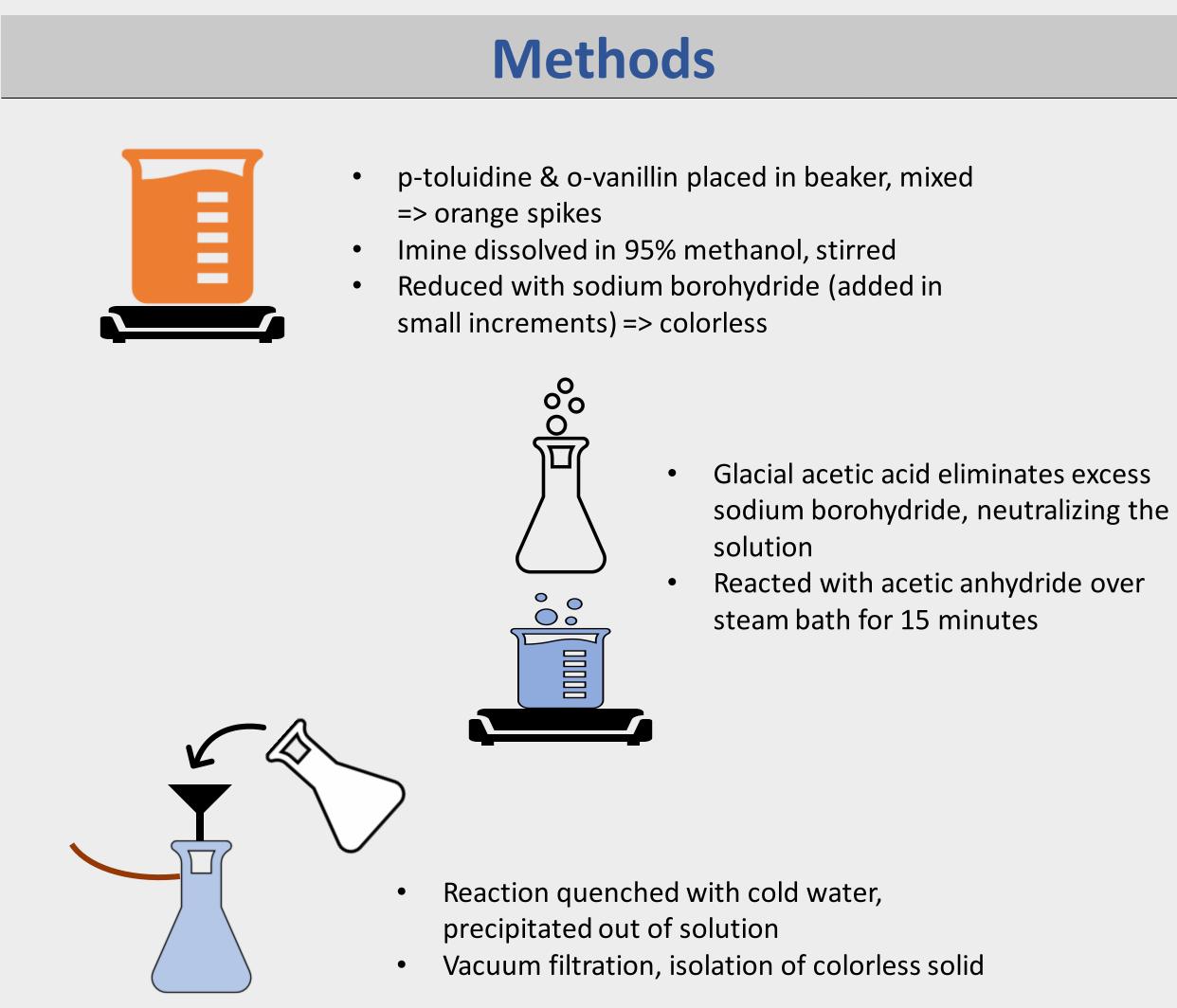
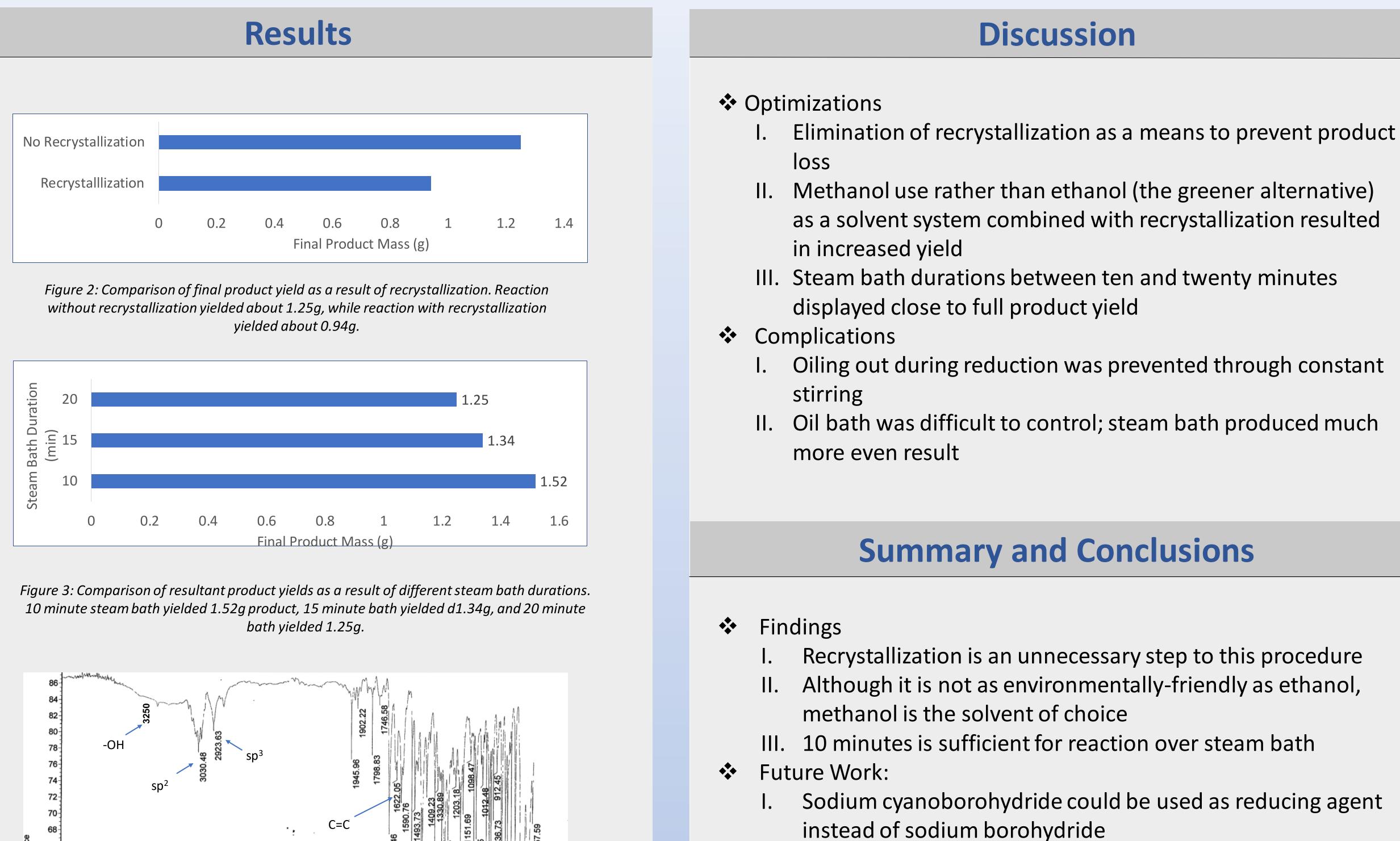
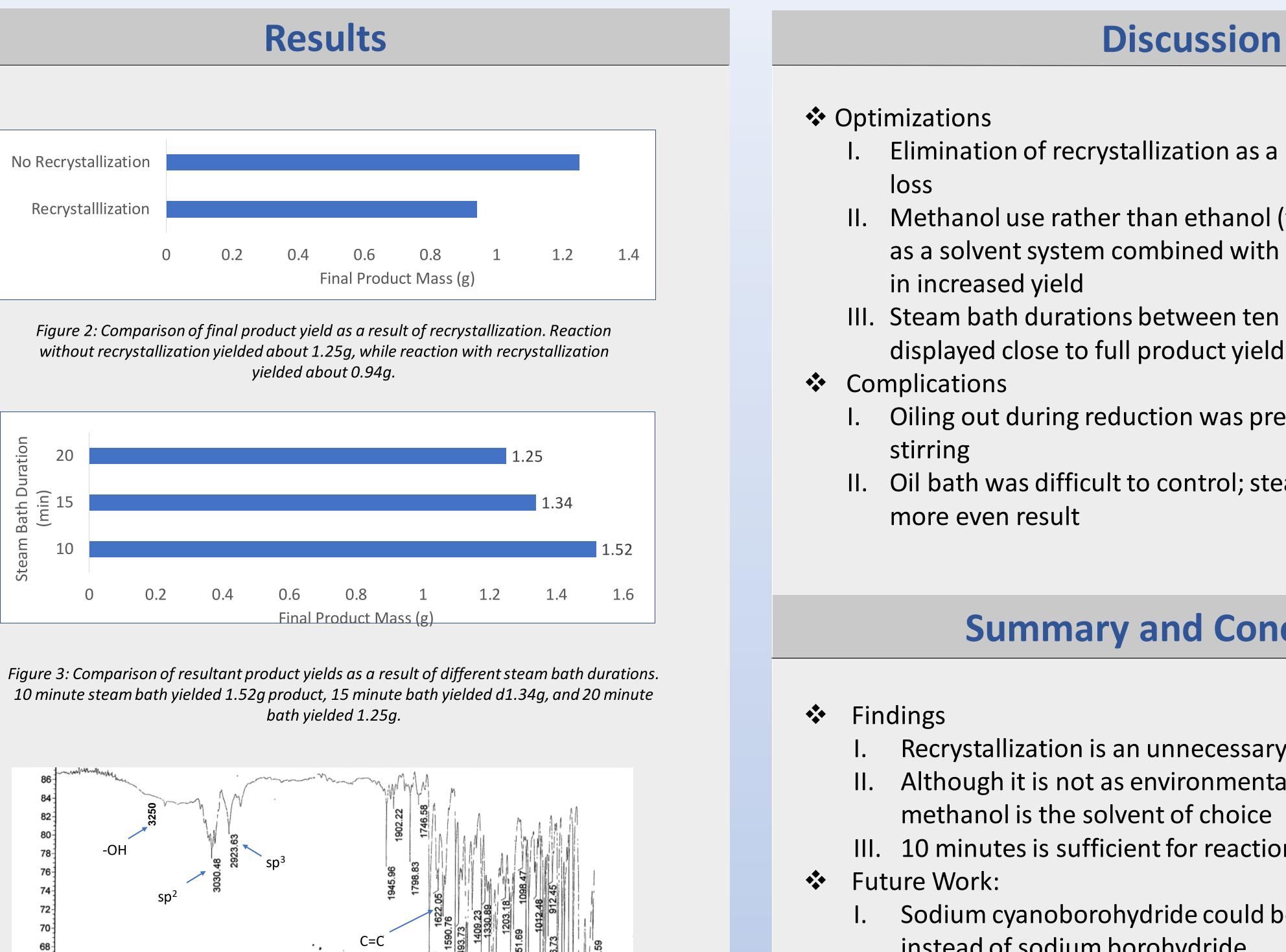


Figure 1. Reductive Amination of N-(2-hydroxy-3-methobenzyl)-N-p-tolyacetamide.

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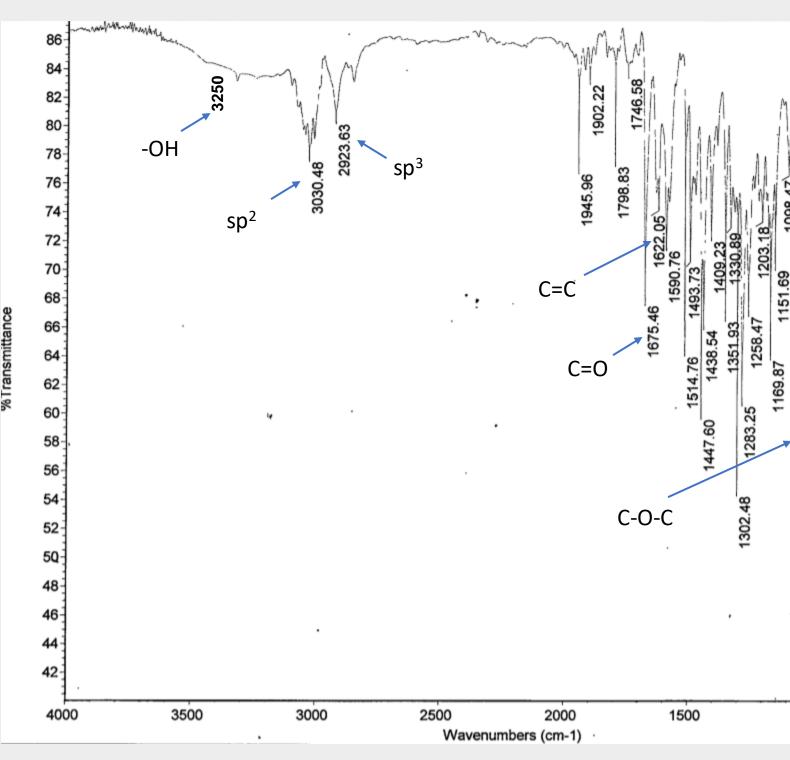


Figure 4: IR spectrum of product N-(2-hydroxy-3-methoxybenzyl)-N-p-tolylacetamide.

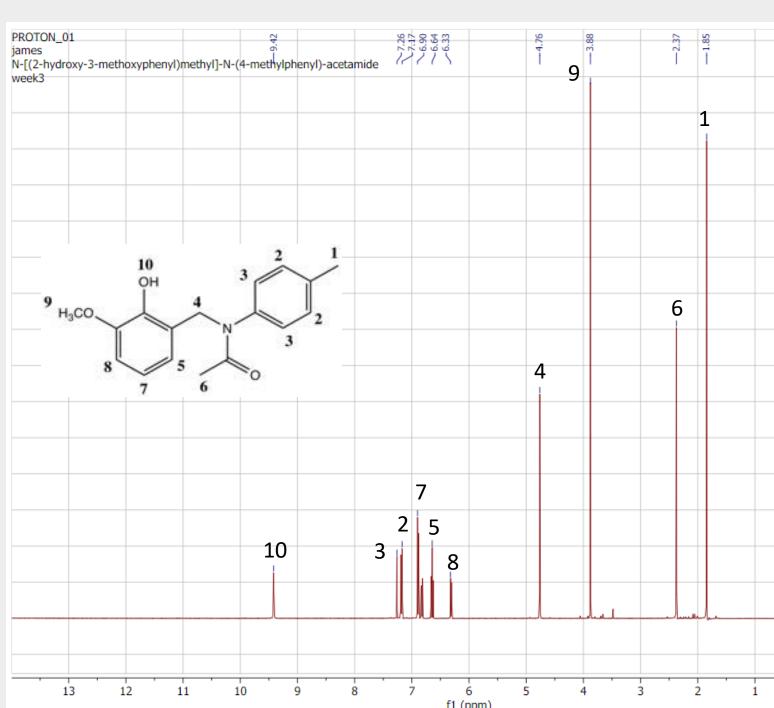


Figure 5. ¹H NMR spectrum of product N-(2-hydroxy-3-methoxybenzyl)-N-p-tolylacetamide.

- II. An acid catalyst could be used to facilitate the formation of the imine
- III. Recrystallization at the end of the filtration step to help further purify product

Acknowledgments

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